# Physics 403 Problem Set 1 

Model Solution

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## Problem 1. Discrete Probabilities

(a) Suppose you throw eight equally weighted six-sided dice. What is the probability that I will get "1" showing up twice, " 3 " showing up once, " 4 " showing up twice, " 5 " showing up twice, and " 6 " showing up once?

Solution: We shall denote the outcome of a roll using the list $r=\left(r_{1}, r_{2}, r_{3}, r_{4}, r_{5}, r_{6}, r_{7}, r_{8}\right)$, where $r_{i}$ is the value rolled by the $i^{\text {th }}$ die. Notice that if two entries $r_{i}$ and $r_{j}$ in the list are equal, then exchanging them in the list leaves $r$ unchanged. This tells us that the number of distinct permuatations is given by the multinomial coefficient $C_{n_{1} \ldots n_{6}}^{8}$ where $n_{i}$ is the number of times that $i$ is rolled. The number of distinct permutations of the roll $r=(1,1,3,4,4,5,5,6)$ is then:

$$
\begin{equation*}
C_{201221}^{8}=\frac{8!}{2!0!1!2!2!1!}=5040 \tag{1.1}
\end{equation*}
$$

Thus, there are 5040 distinct ways to roll the combination of numbers. As each roll configuration is equally probable, the probability of rolling a particular combination is the number of ways to roll that combination divided by the total number of rolls. Each die can have 6 outcomes, and the value on each die is independent of the others. The total number of outcomes is subsequently $6^{8}=1679616$.

We conclude that the probability $P_{r}$ of rolling the listed combination of numbers is $5040 / 1679616$, which we can write as the percentage:

$$
\begin{equation*}
P_{r}=0.3 \% \tag{1.2}
\end{equation*}
$$

(b) Suppose you are dealt a hand of five cards from a deck of cards. What is the probability that you will get exactly three cards of the same value?

First, we shall determine the number of threes of a kind. For a given value, a three of a kind must include three suits of that value and exclude one. We have four choices of suits and must choose three, and the number of ways to do this is $\binom{4}{3}=4$. There are 13 values with which we can make threes of a kind, so there are 52 total threes of a kind.

Next, we consider the number of ways we could draw the other two cards. Out of the 49 remaining cards in the deck, one is the same as the three of a kind drawn. Then, we must draw 2 out of 48 cards. The number
of ways we can do this is given by the binomial coefficient $\binom{48}{2}=1128$. Combined with our previous result, we see that there are $52 \times 1128=58656$ hands with three of a kind.

The probability of drawing three of a kind is equal to the number of ways to draw three of a kind divided by the total number of possible hands. We are drawing 5 cards out of 52 possible cards, which means there are $\binom{52}{5}=2598960$ hands we can draw. Then, the probability $P_{3}$ of drawing three cards of the same value is $58656 / 2598960$, which, as a percentage, is:

$$
\begin{equation*}
P_{3}=2.257 \% \tag{1.3}
\end{equation*}
$$

(c) Suppose you get exactly three of a kind on the first draw, and you discard the two other cards. You then draw two more cards to replace these. What is the probability that these two other cards will be a pair? What is the probability that one of these two other cards will have the same value as the three of a kind you already have?

There are two cases we must consider. First, we consider the case that the two discarded cards were a pair. We shall now count the number of ways we can draw a pair. There is one pair of cards with value of the discarded pair, and no pairs with value of the three of a kind. There are 11 other values for which all four suits remain, and for each of these values, there are $\binom{4}{2}=6$ pairs. Thus, we have $6 \times 11+1=67$ pairs this way. As there are $\binom{47}{2}=1081$ ways we can draw two cards, the probability $C_{1}$ of drawing a pair is $67 / 1081=0.062$.

Now we consider the case that the discarded cards were not a pair. In this case, there are 10 values with all four suits remaining. The number of pairs of these cards that can be drawn is $10 \times\binom{ 4}{2}=60$. There are 2 values for which three suits remain. The number of pairs of these cards that can be drawn is $2 \times\binom{ 3}{2}=6$. The remaining suit cannot have any pairs of it drawn. Once more, the way we can draw two cards is 1081, so the probability $C_{2}$ of drawing a pair is $66 / 1081=0.0611$.

Now, we must consider the probability of each of these scenarios happening. Let $P_{1}$ be the probability of drawing three of a kind with the other two cards being a pair (the first case) and $P_{2}$ be the probability of drawing three of a kind with the other two cards not being a pair (the second case). Then, as we know that we start with a three of a kind, the probability of starting in case one is $P_{1} /\left(P_{1}+P_{2}\right)$, and the probability that we started in case 1 and drew a pair is $C_{1} P_{1} /\left(P_{1}+P_{2}\right)$. Similarly, the probability of starting in case two is $P_{2} /\left(P_{1}+P_{2}\right)$ and the probability of case two happening is $C_{2} P_{2} /\left(P_{1}+P_{2}\right)$. Then, the total probability $P_{32}$ of starting with a three of a kind, discarding two cards, and drawing a pair is:

$$
\begin{equation*}
P_{32}=\frac{C_{1} P_{1}}{P_{1}+P_{2}}+\frac{C_{2} P_{2}}{P_{1}+P_{2}} \tag{1.4}
\end{equation*}
$$

We must now determine $P_{1}$ and $P_{2}$. We shall start with $P_{1}$. If three of the cards have the same value, then the other pair could be one of 12 values. Each of these values has four suits, so there are $12 \times\binom{ 4}{2}=72$ ways that these cards could be a pair. Then, recalling from the previous part that there are 52 threes of a kind, the probability of drawing five cards, having three be of one value and two be of another, is $72 \times 52 /\binom{52}{5}=0.00144$.

Next, we shall determine $P_{2}$. If three of the cards have the same value, then the two remaining cards must be two distinct values out of the remaining 12 . There are $\binom{12}{2}=66$ ways that these can be drawn. There are 4 suits for each of these values, so there are $66 \times 4 \times 4=1056$ ways to draw these pairs. There are 52 ways to draw a three of a kind, so the probability of drawing three of a kind without a pair is $52 \times 1056 /\binom{52}{5}=0.0211$. Substituting this into our expression for $P_{32}$, we find:

$$
\begin{equation*}
P_{32}=\frac{0.00144 \times 0.062}{0.0211+0.00144}+\frac{0.0211 \times 0.0611}{0.0211+0.00144} \tag{1.5}
\end{equation*}
$$

We can now evaluate the right hand side to the following percentage:

$$
\begin{equation*}
P_{32}=6.12 \% \tag{1.6}
\end{equation*}
$$

We shall now determine the probability of drawing a card with the same value as the three of a kind. There is one card left in the deck that satisfies this criterion, so one of the cards must be this card. There are 46 other cards we can draw, which leaves us with 46 possible draws to obtain a four of a kind. There are $\binom{47}{2}=1081$ total pairs we can draw, so the probability $P_{4}$ of obtaining the fourth card is $46 / 1081$, which, in percentage form, we can write as:

$$
\begin{equation*}
P_{4}=4.26 \% \tag{1.7}
\end{equation*}
$$

## Problem 2. Thermodynamics for a Magnetic System

Suppose we have a magnetic system with the equation of state:

$$
\begin{equation*}
M(T, B)=\frac{C B}{T} \tag{2.1}
\end{equation*}
$$

where $M$ is the magnetization, $B$ is the magnetic field, $T$ is a temperature, and $C$ is some constant. The energy $U$ of this system is:

$$
\begin{equation*}
U=-M B \tag{2.2}
\end{equation*}
$$

If the field $B$ is changed by an amount $d B$, the work $d W$ done by the system is given by:

$$
\begin{equation*}
d W=M d B \tag{2.3}
\end{equation*}
$$

(a) Show that the heat given to the system under simultaneous changes $d B$ and $d M$ satisfies $d Q=-B d M$.

The first law of thermodynamics tells us that:

$$
\begin{equation*}
d U=d Q-d W \tag{2.4}
\end{equation*}
$$

Writing $d W$ in terms of the magnetization and the magnetic field, we find:

$$
\begin{equation*}
d U=d Q-M d B \tag{2.5}
\end{equation*}
$$

Next, using the product rule, we can differentiate our expression for the internal energy $U$ to find that:

$$
\begin{equation*}
d U=-M d B-B d M \tag{2.6}
\end{equation*}
$$

Substituting this into the left hand side of our previous expression and solving for $d Q$, yields:

$$
\begin{equation*}
d Q=-B d M \tag{2.7}
\end{equation*}
$$

(b) Using the previous result, find the differential change in the entropy $d S$ and the form $S(M)$ for the entropy of the system.

The differential $d S$ of the entropy is given by:

$$
\begin{equation*}
d S=\frac{d Q}{T} \tag{2.8}
\end{equation*}
$$

Substituting in our expression from the previous part, we find:

$$
\begin{equation*}
d S=-\frac{B}{T} d M \tag{2.9}
\end{equation*}
$$

Now, we can rearrange the equation of state for the system to find:

$$
\begin{equation*}
B=\frac{M T}{C} \tag{2.10}
\end{equation*}
$$

Substituting this into our expression for the entropy, we have:

$$
\begin{equation*}
d S=-\frac{M}{C} d M \tag{2.11}
\end{equation*}
$$

Finally, integrating both sides, obtain the following expression for the entropy:

$$
\begin{equation*}
S(M)=S(0)-\frac{M^{2}}{2 C} \tag{2.12}
\end{equation*}
$$

## Problem 3. N Spin-1/2 Systems

Consider a set of $N$ non-interacting spin-1/2 systems in a magnetic field, such that the energies of each of the individual spins are $E_{1}$ and $E_{2}$ respectively.
(a) Find the partition function for this system. Find the average energy $U$ of the total system at a temperature $T$. From this expression, derive the specific heat $C_{V}$.

We recognize that the particle number is being held constant, so we considering the canocical ensemble. Then, the partition function $Z$ is given by:

$$
\begin{equation*}
Z=\sum_{\text {states }} e^{-\beta E_{s}} \tag{2.1}
\end{equation*}
$$

where $E_{s}$ is th energy of the state. For a state $s$, let $s_{1}$ and $s_{2}$ denote the number of spin systems in the first state and second state respectively. Then, the energy $E_{s}$ of the state is:

$$
\begin{equation*}
E_{s}=s_{1} E_{1}+s_{2} E_{2} \tag{2.2}
\end{equation*}
$$

Notice that the spin systems must either be in the $s_{1}$ or the $s_{2}$ spin state, so we have the constraint that:

$$
\begin{equation*}
N=s_{1}+s_{2} \tag{2.3}
\end{equation*}
$$

Then, our energy $E_{s}$ can be written in the following way:

$$
\begin{equation*}
E_{s}=s_{1} E_{1}+\left(N-s_{1}\right) E_{2} \tag{2.4}
\end{equation*}
$$

Substituting this into our expression for the partition function, we find:

$$
\begin{equation*}
Z=\sum_{\text {states }} e^{-\beta s_{1} E_{1}} e^{-\beta\left(N-s_{1}\right) E_{2}} \tag{2.5}
\end{equation*}
$$

We have written the energy of the state such that it only depends on $s_{1}$, so we can express the sum over states as a sum over the allowed values of $s_{1}$. The minimum allowed value of $s_{1}$ is zero, and the maximum allowed value is $N$. Then:

$$
\begin{equation*}
Z=\sum_{s_{1}=0}^{N} g\left(s_{1}\right) e^{-\beta s_{1} E_{1}} e^{-\beta\left(N-s_{1}\right) E_{2}} \tag{2.6}
\end{equation*}
$$

where $g$ is the degeneracy of states for each value of $s_{1}$. The number of ways that we can have $s_{1}$ systems in the first spin state is the number of ways that we can choose $s_{1}$ systems out of the $N$ total systems, so it is equal to the binomial coefficient $\binom{N}{s_{1}}$. This tells us that:

$$
\begin{equation*}
Z=\sum_{s_{1}=0}^{N}\binom{N}{s_{1}} e^{-\beta s_{1} E_{1}} e^{-\beta\left(N-s_{1}\right) E_{2}} \tag{2.7}
\end{equation*}
$$

The binomial theorem tells us that the right hand side of this expression is the series expansion of a binomial in $e^{-\beta E_{1}}$ and $e^{-\beta E_{2}}$. The series subsequently evaluates to the following:

$$
\begin{equation*}
Z=\left(e^{-\beta E_{1}}+e^{-\beta E_{2}}\right)^{N} \tag{2.8}
\end{equation*}
$$

Note: This is a very important result. The preceding argument can be generalized to tell us that the partition function of a system made up of non-interacting subsystems is equal to the product of the partition function of the subsystems.

The average energy $U$ of the system is given by:

$$
\begin{equation*}
U=-\frac{1}{Z} \frac{\partial Z}{\partial \beta} \tag{2.9}
\end{equation*}
$$

Substituting in our expression for the partition function:

$$
\begin{equation*}
U=-\frac{1}{\left(e^{-\beta E_{1}}+e^{-\beta E_{2}}\right)^{N}} \frac{\partial}{\partial \beta}\left(e^{-\beta E_{1}}+e^{-\beta E_{2}}\right)^{N} \tag{2.10}
\end{equation*}
$$

Next, we evaluate the derivative and simplify, which yields:

$$
\begin{equation*}
U=\frac{N\left(E_{1} e^{-E_{1} / k_{B} T}+E_{2} e^{-E_{2} / k_{B} T}\right)}{e^{-E_{1} / k_{B} T}+e^{-E_{2} / k_{B} T}} \tag{2.11}
\end{equation*}
$$

Does this make sense? This is the result we would naively expect, as it is just the probability weighted average of the energies.

Next, we shall calculate the specific heat at constant volume $C_{V}$. By definition, we know that:

$$
\begin{equation*}
C_{V}(T)=\left(\frac{\partial U}{\partial T}\right)_{V} \tag{2.12}
\end{equation*}
$$

In terms of $\beta$, we can write this as:

$$
\begin{equation*}
C_{V}(T)=-\frac{1}{k_{B} T^{2}}\left(\frac{\partial U}{\partial \beta}\right)_{V} \tag{2.13}
\end{equation*}
$$

Substituting in our expression for the internal energy, we find:

$$
\begin{equation*}
C_{V}(T)=-\frac{N}{k_{B} T^{2}} \frac{\partial}{\partial \beta} \frac{E_{1} e^{-\beta E_{1}}+E_{2} e^{-\beta E_{2}}}{e^{-\beta E_{1}}+e^{-\beta E_{2}}} \tag{2.14}
\end{equation*}
$$

We can now use the quotient rule to evaluate the derivative:

$$
\begin{equation*}
C_{V}(T)=\frac{N\left[\left(E_{1} e^{-\beta E_{1}}+E_{2} e^{-\beta E_{2}}\right)^{2}-\left(E_{1}^{2} e^{-\beta E_{1}}+E_{2}^{2} e^{-\beta E_{2}}\right)\left(e^{-\beta E_{1}}+e^{-\beta E_{2}}\right)\right]}{k_{B} T^{2}\left(e^{-\beta E_{1}}+e^{-\beta E_{2}}\right)^{2}} \tag{2.15}
\end{equation*}
$$

We can now simplify this expression to obtain the following:

$$
\begin{equation*}
C_{V}(T)=\frac{N\left(E_{1}-E_{2}\right)^{2}}{k_{B} T^{2}\left(e^{\left(E_{1}-E_{2}\right) / 2 k_{B} T}+e^{\left(E_{2}-E_{1}\right) / 2 k_{B} T}\right)^{2}} \tag{2.16}
\end{equation*}
$$

(b) Find the expressions for $U$ and $C_{V}$ in the limit where $T$ is very large. That is, find the limit as $T$ goes to infinity along with the first nonvanishing correction for very large but finite $T$.

To find the limiting behavior of the expressions obtained in the previous part, it is useful to write them as hyperbolic trigonometric functions. We can rewrite our expression for $U$ as:

$$
\begin{equation*}
U=\frac{N\left(E_{1} e^{\beta\left(E_{2}-E_{1}\right) / 2}+E_{2} e^{\beta\left(E_{1}-E_{2}\right) / 2}\right)}{e^{-\beta\left(E_{1}-E_{2}\right) / 2}+e^{-\beta\left(E_{2}-E_{1}\right) / 2}} \tag{2.17}
\end{equation*}
$$

Comparing the denominator to the definition of the hyperbolic cosine, we see that:

$$
\begin{align*}
U=\frac{N}{2} \operatorname{sech}\left(\frac{\beta\left(E_{1}-E_{2}\right)}{2}\right)\left(\frac{E_{1}+E_{2}}{2}\right. & e^{-\beta\left(E_{1}-E_{2}\right) / 2}+\frac{E_{1}-E_{2}}{2} e^{-\beta\left(E_{1}-E_{2}\right) / 2}  \tag{2.18}\\
& \left.+\frac{E_{1}+E_{2}}{2} e^{\beta\left(E_{1}-E_{2}\right) / 2}+\frac{E_{2}-E_{1}}{2} e^{\beta\left(E_{1}-E_{2}\right) / 2}\right)
\end{align*}
$$

Next, we can write the quantities inside the parentheses in terms of hyperbolic trigonometric functions, which yields:

$$
\begin{equation*}
U=\frac{N\left(E_{1}+E_{2}\right)}{2}+\frac{N\left(E_{1}-E_{2}\right)}{2} \tanh \left(\frac{\beta\left(E_{2}-E_{1}\right)}{2}\right) \tag{2.19}
\end{equation*}
$$

We now expand this expression about $\beta=0$, which, to leading order, gives us the equation:

$$
\begin{equation*}
U \approx \frac{N\left(E_{1}+E_{2}\right)}{2}-\frac{N\left(E_{1}-E_{2}\right)^{2}}{4} \operatorname{sech}^{2}(0) \beta \tag{2.20}
\end{equation*}
$$

Thus, we conclude that in the large temperature regime, the average energy is given by:

$$
\begin{equation*}
U(T) \approx \frac{N\left(E_{1}+E_{2}\right)}{2}-\frac{N\left(E_{1}-E_{2}\right)^{2}}{4 k_{B} T} \tag{2.21}
\end{equation*}
$$

Does this make sense? In the inifinite temperature limit, there is an equal probability of being in either of the two states. Thus, the leading order term must be the average of the two energies multiplied by the number of particles. In the finite temperature regime, there is a higher probability of particles being in a lower energy state, with a weight that only depends on the energy difference between the two states, which results in a negative correction to the leading order term. Notice further that this correction will be the same regardless of whether $E_{1}$ or $E_{2}$ has a greater energy, so it must be invariant under their exchange. This, combined with the fact that the term must have dimensions of energy means that it must be quadratic in the energy difference.

We shall now perform a similar procedure with the specific heat. Rewriting our expression from the previous part in terms of hyperbolic trigonometric functions, we have:

$$
\begin{equation*}
C_{V}(T)=\frac{N k_{B}\left(E_{1}-E_{2}\right)^{2} \beta^{2}}{4} \operatorname{sech}^{2}\left(\frac{\beta\left(E_{1}-E_{2}\right)}{2}\right) \tag{2.22}
\end{equation*}
$$

Firt, we recognize that as $\beta$ goes to zero, this expression vanishes. Notice that the only dependence on $\beta$ that is not in the form of a polynomial comes from the hyperbolic secant. Then, we only need to expand that term about $\beta=0$. Notice that:

$$
\begin{equation*}
\operatorname{sech} x=\frac{2}{e^{-x}+e^{x}} \approx \frac{2}{2+2 x+\mathcal{O}\left(x^{2}\right)}=1+\mathcal{O}(x) \tag{2.23}
\end{equation*}
$$

Then, to lowest non-trivial order, we have:

$$
\begin{equation*}
C_{V}(T) \approx \frac{N k_{B}\left(E_{1}-E_{2}\right)^{2}}{4 k_{B} T^{2}} \tag{2.24}
\end{equation*}
$$

Does this make sense? In the inifinite temperature limit, an increase in temperature does not affect the state populations, and subsequently does not alter the energy of the system. Thus, the leading order term must vanish. In the finite temperature regime an increase in temperature will change the probabilities of state population according to the difference in the energies. Using a similar argument as with the average energy, we see that the first non-trivial term must depend on the energies as some function of $\left(E_{1}-E_{2}\right)^{2}$.

